

# Smart ZMC plus Moly Sulphate 6.1-5.1-1.5

# **Crop Smart Pty Ltd**

Chemwatch: 5669-16 Version No: 2.1

Safety Data Sheet according to Work Health and Safety Regulations (Hazardous Chemicals) 2023 and ADG requirements

Issue Date: 28/03/2024 Print Date: 04/04/2024

S.GHS.AUS.EN.E

#### SECTION 1 Identification of the substance / mixture and of the company / undertaking

#### **Product Identifier**

Product name	Smart ZMC plus Moly 6.1-5.1-1.5	
Chemical Name	Not Applicable	
Synonyms	Not Available	
Proper shipping name	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)	
Chemical formula	Not Applicable	
Other means of identification	Not Available	

# Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Foliar Fertiliser for application in agriculture.
Relevant luentineu uses	Use according to manufacturer's directions.

#### Details of the manufacturer or supplier of the safety data sheet

Registered company name	Crop Smart Pty Ltd	
Registered company name		
Address	2409/ 4 Daydream Street WARRIEWOOD NSW 2102 Australia	
Telephone	+61 1300 783 481	
Fax	Not Available	
Website	www.cropsmart.com.au	
Email	Compliance@cropsmart.com.au	

#### Emergency telephone number

Association / Organisation	CHEMWATCH EMERGENCY RESPONSE (24/7)	
Emergency telephone numbers	+61 1800 951 288	
Other emergency telephone numbers	+61 3 9573 3188	

Once connected and if the message is not in your preferred language then please dial 01

#### **SECTION 2 Hazards identification**

# Classification of the substance or mixture

Poisons Schedule	S6
Classification <sup>[1]</sup>	Acute Toxicity (Oral) Category 4, Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 1, Acute Toxicity (Inhalation) Category 4, Specific Target Organ Toxicity - Repeated Exposure Category 2, Hazardous to the Aquatic Environment Long-Term Hazard Category 1
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

#### Label elements

Signal word

Danger

# Hazard statement(s)

H302	Harmful if swallowed.
H315	Causes skin irritation.
H318	Causes serious eye damage.
H332	Harmful if inhaled.

H373	May cause damage to organs through prolonged or repeated exposure.	
H410	Very toxic to aquatic life with long lasting effects.	

# Precautionary statement(s) Prevention

P260	Do not breathe mist/vapours/spray.	
P271	Use only outdoors or in a well-ventilated area.	
P280	Wear protective gloves, protective clothing, eye protection and face protection.	
P264	Wash all exposed external body areas thoroughly after handling.	
P270	Do not eat, drink or smoke when using this product.	
P273	Avoid release to the environment.	

# Precautionary statement(s) Response

P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P310	Immediately call a POISON CENTER/doctor/physician/first aider.	
P391	Collect spillage.	
P301+P312	IF SWALLOWED: Call a POISON CENTER/doctor/physician/first aider if you feel unwell.	
P302+P352	IF ON SKIN: Wash with plenty of water.	
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.	
P330	Rinse mouth.	
P332+P313	If skin irritation occurs: Get medical advice/attention.	
P362+P364	Take off contaminated clothing and wash it before reuse.	

## Precautionary statement(s) Storage

Not Applicable

Precautionary statement(s) Disposal

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

## **SECTION 3 Composition / information on ingredients**

P501

#### Substances

See section below for composition of Mixtures

# Mixtures

CAS No	%[weight]	Name
10034-96-5	10-30	manganese sulfate, hydrate
7446-20-0	10-30	zinc sulfate heptahydrate
7758-99-8	1-10	copper sulfate, pentahydrate
10102-40-6	<1	sodium molybdate
7664-38-2	<1	phosphoric acid
Legend:	1. Classified by Chernwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available	

#### **SECTION 4 First aid measures**

## Description of first aid measures

Eye Contact	<ul> <li>If this product comes in contact with the eyes:</li> <li>Immediately hold eyelids apart and flush the eye continuously with running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.</li> <li>Transport to hospital or doctor without delay.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>	
Skin Contact	<ul> <li>If skin or hair contact occurs:</li> <li>Immediately flush body and clothes with large amounts of water, using safety shower if available.</li> <li>Quickly remove all contaminated clothing, including footwear.</li> <li>Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.</li> <li>Transport to hospital, or doctor.</li> </ul>	
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor, without delay.</li> </ul>	
Ingestion	<ul> <li>For advice, contact a Poisons Information Centre or a doctor at once.</li> <li>Urgent hospital treatment is likely to be needed.</li> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casuality can comfortably drink.</li> <li>Transport to hospital or doctor without delay.</li> </ul>	

#### Indication of any immediate medical attention and special treatment needed

#### for copper intoxication:

Unless extensive vomiting has occurred empty the stomach by lavage with water, milk, sodium bicarbonate solution or a 0.1% solution of potassium ferrocyanide (the resulting copper ferrocyanide is insoluble).

- Administer egg white and other demulcents.
- Maintain electrolyte and fluid balances.
- Morphine or meperidine (Demerol) may be necessary for control of pain.
- If symptoms persist or intensify (especially circulatory collapse or cerebral disturbances, try BAL intramuscularly or penicillamine in accordance with the supplier's recommendations.
- Treat shock vigorously with blood transfusions and perhaps vasopressor amines.
- If intravascular haemolysis becomes evident protect the kidneys by maintaining a diuresis with mannitol and perhaps by alkalinising the urine with sodium bicarbonate.
- It is unlikely that methylene blue would be effective against the occassional methaemoglobinemia and it might exacerbate the subsequent haemolytic episode.

Institute measures for impending renal and hepatic failure.
 [GOSSELIN, SMITH & HODGE: Commercial Toxicology of Commercial Products]

A role for activated charcoals for emesis is, as yet, unproven.

- In severe poisoning CaNa2EDTA has been proposed.
- [ELLENHORN & BARCELOUX: Medical Toxicology]

Both dermal and oral toxicity of manganese salts is low because of limited solubility of manganese. No known permanent pulmonary sequelae develop after acute manganese exposure. Treatment is supportive.

#### [Ellenhorn and Barceloux: Medical Toxicology]

In clinical trials with miners exposed to manganese-containing dusts, L-dopa relieved extrapyramidal symptoms of both hypo kinetic and dystonic patients. For short periods of time symptoms could also be controlled with scopolamine and amphetamine. BAL and calcium EDTA prove ineffective.

[Gosselin et al: Clinical Toxicology of Commercial Products.]

## **SECTION 5 Firefighting measures**

#### Extinguishing media

- Foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.
- Water spray or fog Large fires only.

## Special hazards arising from the substrate or mixture

 Fire Incompatibility

 Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

 Advice for firefighters

 Alert Fire Brigade and tell them location and nature of hazard.
 Wear full body protective clothing with breathing apparatus.
 Prevent, by any means available, spillage from entering drains or water course.
 Use water delivered as a fine sorav to control fire and cool adiacent area.

HAZCHEM	•3Z
Fire/Explosion Hazard	<ul> <li>Combustible.</li> <li>Slight fire hazard when exposed to heat or flame.</li> <li>Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>On combustion, may emit toxic fumes of carbon monoxide (CO).</li> <li>May emit acrid smoke.</li> <li>Mists containing combustible materials may be explosive.</li> <li>Combustion products include:</li> <li>carbon dioxide (CO2)</li> <li>sulfur oxides (SOx)</li> <li>sulfur dioxide (SO2)</li> <li>metal oxides</li> <li>other pyrolysis products typical of burning organic material.</li> </ul>
Fire Fighting	<ul> <li>Use water delivered as a fine spray to control fire and cool adjacent area.</li> <li>Avoid spraying water onto liquid pools.</li> <li>DO NOT approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> </ul>

## **SECTION 6 Accidental release measures**

# Personal precautions, protective equipment and emergency procedures

See section 8

## Environmental precautions

See section 12

## Methods and material for containment and cleaning up

Minor Spills	<ul> <li>Environmental hazard - contain spillage.</li> <li>Clean up all spills immediately.</li> <li>Avoid breathing vapours and contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Contain and absorb spill with sand, earth, inert material or vermiculite.</li> <li>Wipe up.</li> <li>Place in a suitable, labelled container for waste disposal.</li> </ul>
Major Spills	<ul> <li>Environmental hazard - contain spillage.</li> <li>Moderate hazard.</li> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear breathing apparatus plus protective gloves.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> </ul>

<ul> <li>Contain spill with sand, earth or vermiculite.</li> <li>Collect recoverable product into labelled containers for recycling.</li> <li>Absorb remaining product with sand, earth or vermiculite.</li> <li>Collect solid residues and seal in labelled drums for disposal.</li> </ul>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

## **SECTION 7 Handling and storage**

Safe handling	<ul> <li>DO NOT allow clothing wet with material to stay in contact with skin</li> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> <li>DO NOT enter confined spaces until atmosphere has been checked.</li> <li>Avoid smoking, naked lights or ignition sources.</li> <li>Avoid contact with incompatible materials.</li> <li>When handling, DO NOT eat, drink or smoke.</li> <li>Keep containers securely sealed when not in use.</li> <li>Avoid physical damage to containers.</li> <li>Always wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately.</li> <li>Use good occupational work practice.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions.</li> </ul>
Other information	<ul> <li>Store in original containers.</li> <li>Keep containers securely sealed.</li> <li>No smoking, naked lights or ignition sources.</li> <li>Store in a cool, dry, well-ventilated area.</li> <li>Store away from incompatible materials and foodstuff containers.</li> <li>Protect containers against physical damage and check regularly for leaks.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> </ul>

# Conditions for safe storage, including any incompatibilities

Suitable container	<ul> <li>Metal can or drum</li> <li>Packaging as recommended by manufacturer.</li> <li>Check all containers are clearly labelled and free from leaks.</li> </ul>
Storage incompatibility	Avoid reaction with oxidising agents

## **SECTION 8 Exposure controls / personal protection**

## **Control parameters**

## Occupational Exposure Limits (OEL)

## INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	manganese sulfate, hydrate	Manganese, dust & compounds (as Mn)	1 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	sodium molybdate	Molybdenum, soluble compounds (as Mo)	5 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	phosphoric acid	Phosphoric acid	1 mg/m3	3 mg/m3	Not Available	Not Available

Emergency Limits

Ingredient	TEEL-1 TEEL-2			TEEL-3
manganese sulfate, hydrate	9.2 mg/m3	15 mg/m3		90 mg/m3
manganese sulfate, hydrate	8.2 mg/m3	14 mg/m3		430 mg/m3
zinc sulfate heptahydrate	27 mg/m3	170 mg/m3		1,000 mg/m3
zinc sulfate heptahydrate	15 mg/m3	97 mg/m3		580 mg/m3
copper sulfate, pentahydrate	7.5 mg/m3 9.9 mg/m3			59 mg/m3
copper sulfate, pentahydrate	12 mg/m3	32 mg/m3		190 mg/m3
sodium molybdate	3.8 mg/m3	34 mg/m3		210 mg/m3
sodium molybdate	3.2 mg/m3	17 mg/m3		100 mg/m3
phosphoric acid	Not Available	Not Available		Not Available
Ingredient	Original IDLH	Original IDLH		
manganese sulfate, hydrate	500 mg/m3		Not Available	
zinc sulfate heptahydrate	Not Available		Not Available	
copper sulfate, pentahydrate	Not Available		Not Available	
sodium molybdate	1,000 mg/m3		Not Available	
phosphoric acid	1,000 mg/m3		Not Available	

Occupational Exposure Banding

inc sulfate heptahydrate	E E Occupational exposure banding is a process of assigning che adverse health outcomes associated with exposure. The out to a range of exposure concentrations that are expected to pr Engineering controls are used to remove a hazard or place a can be highly effective in protecting workers and will typically	ut of this process is an occupational exposure band (OE	
lotes:	Occupational exposure banding is a process of assigning che adverse health outcomes associated with exposure. The output to a range of exposure concentrations that are expected to pro- Engineering controls are used to remove a hazard or place a can be highly effective in protecting workers and will typically	micals into specific categories or bands based on a cher ut of this process is an occupational exposure band (OE	
	adverse health outcomes associated with exposure. The output to a range of exposure concentrations that are expected to provide the provide the provide the provide the provide the provide the provided	ut of this process is an occupational exposure band (OE	
posure controls	can be highly effective in protecting workers and will typically		
	can be highly effective in protecting workers and will typically		
Appropriate engineering controls	The basic types of engineering controls are: Process controls which involve changing the way a job activit Enclosure and/or isolation of emission source which keeps a strategically "adds" and "removes" air in the work environmer design of a ventilation system must match the particular proce Employers may need to use multiple types of controls to prev Local exhaust ventilation usually required. If risk of overexpose protection. Supplied-air type respirator may be required in spe An approved self contained breathing apparatus (SCBA) may Provide adequate ventilation in warehouse or closed storage velocities which, in turn, determine the "capture velocities" of Type of Contaminant: solvent, vapours, degreasing etc., evaporating from tank (ir aerosols, fumes from pouring operations, intermittent conta spray drift, plating acid fumes, pickling (released at low veloc direct spray, spray painting in shallow booths, drum filling, o generation into zone of rapid air motion) grinding, abrasive blasting, tumbling, high speed wheel gen of very high rapid air motion). Within each range the appropriate value depends on: Lower end of the range 1: Room air currents minimal or favourable to capture 2: Contaminants of low toxicity or of nuisance value only. 3: Intermittent, low production. 4: Large hood or large air mass in motion Simple theory shows that air velocity falls rapidly with distance decreases with the square of distance from the extraction poi adjusted, accordingly, after reference to distance from the cor	be independent of worker interactions to provide this high y or process is done to reduce the risk. selected hazard "physically" away from the worker and v t. Ventilation can remove or dilute an air contaminant if d ass and chemical or contaminant in use. ent employee overexposure. sure exists, wear approved respirator. Correct fit is essent acial circumstances. Correct fit is essential to ensure ade be required in some situations. area. Air contaminants generated in the workplace posse fresh circulating air required to effectively remove the cor- strest exists, wear approved respirator. Correct fit is essential to ensure ade be required in some situations. area. Air contaminants generated in the workplace posse fresh circulating air required to effectively remove the cor- strest dusts (released conveyer transfers, welding, with into zone of active generation) is onveyer loading, crusher dusts, gas discharge (active erated dusts (released at high initial velocity into zone Upper end of the range 1: Disturbing room air currents 2: Contaminants of high toxicity 3: High production, heavy use 4: Small hood-local control only e away from the opening of a simple extraction pipe. Velo t (in simple cases). Therefore the air speed at the extraction taminating source. The air velocity at the extraction fan,	n level of protection. entilation that esigned properly. The tial to obtain adequate quate protection. ess varying "escape" taminant. Air Speed: 0.25-0.5 m/s (50- 100 f/min.) 0.5-1 m/s (100- 200 f/min.) 1-2.5 m/s (200- 500 f/min.) 2.5-10 m/s (500- 2000 f/min.)
Individual protection measures, such as personal protective equipment	a minimum of 1-2 m/s (200-400 f/min) for extraction of solven mechanical considerations, producing performance deficits w multiplied by factors of 10 or more when extraction systems a	ts generated in a tank 2 meters distant from the extractio ithin the extraction apparatus, make it essential that theo	n point. Other
Eye and face protection	<ul> <li>Safety glasses with unperforated side shields may be use are not sufficient where complete eye protection is neede or if the material may be under pressure.</li> <li>Chemical goggles. Whenever there is a danger of the ma 1337.1, EN166 or national equivalent]</li> <li>Full face shield (20 cm, 8 in minimum) may be required for protection.</li> <li>Alternatively a gas mask may replace splash goggles and</li> <li>Contact lenses may pose a special hazard; soft contact le describing the wearing of lenses or restrictions on use, st lens absorption and adsorption for the class of chemicals should be trained in their removal and suitable equipmen- irrigation immediately and remove contact lens as soon a irritation - lens should be removed in a clean environment letellicence Rulletine 60.</li> </ul>	d such as when handling bulk-quantities, where there is terial coming in contact with the eyes; goggles must be p or supplementary but never for primary protection of eyes I face shields. In the should be created for each workplace or task. This should in use and an account of injury experience. Medical and is should be readily available. In the event of chemical exp s practicable. Lens should be removed at the first signs of	a danger of splashing, roperly fitted. [AS/NZS ; these afford face cy document, include a review of first-aid personnel oosure, begin eye of eye redness or
Skin protostica	Intelligence Bulletin 59].		
Skin protection	See Hand protection below		

When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

NOTE:

• The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.

• Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.

The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended. Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

· frequency and duration of contact,

chemical resistance of glove material,

glove thickness and

dexterity

Recommended material(s)	Respiratory protection
Other protection	<ul> <li>Overalls.</li> <li>P.V.C apron.</li> <li>Barrier cream.</li> <li>Skin cleansing cream.</li> <li>Eye wash unit.</li> </ul>
Body protection	See Other protection below
	Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent). • When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended. • When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended. • When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended. • Contaminated gloves should be replaced. As defined in ASTM F-739-96 in any application, gloves are rated as: • Excellent when breakthrough time > 480 min • Good when breakthrough time < 20 min • Fair when breakthrough time < 20 min • Fair when plove material degrades For general applications, gloves with a thickness typically greater than 0.35 mm, are recommended. It should be emphasised that glove thickness is not necessarily a good predictor of glove resistance to a specific chemical, as the permeation efficiency of the glove will be dependent on the exact composition of the glove material. Therefore, glove selection should also be based on consideration of the task requirements and knowledge of breakthrough times. Glove thickness may also vary depending on the glove sof varying thickness may be required for specific ctasks. For example: • Thinner gloves (down to 0.1 mm or less) may be required where a high degree of manual dexterity is needed. • Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index". The effect(s) of the following substance(s) are taken into account in the computer-

generated selection: Smart Sulphate ZMC plus Moly

Material	CPI
NAT+NEOPR+NITRILE	А
NATURAL RUBBER	A
NATURAL+NEOPRENE	А
NEOPRENE	А
NEOPRENE/NATURAL	A
NITRILE	А
NITRILE+PVC	А
PE	A
PVC	А
SARANEX-23	А

CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation.

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

#### Ansell Glove Selection

Glove — In order of recommendation
AlphaTec® Solvex® 37-675
AlphaTec 02-100
AlphaTec® Solvex® 37-185
AlphaTec® 58-008
AlphaTec® 58-530B
AlphaTec® 58-530W
AlphaTec® 58-735
AlphaTec® 79-700
AlphaTec® 38-612
DermaShield™ 73-711

The suggested gloves for use should be confirmed with the glove supplier.

#### Respiratory protection

Type AB-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	AB-AUS P2	-	AB-PAPR-AUS / Class 1 P2
up to 50 x ES	-	AB-AUS / Class 1 P2	-
up to 100 x ES	-	AB-2 P2	AB-PAPR-2 P2 ^

- Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

· The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option). Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government

mandated or vendor recommended. · Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory

protection program. Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU)

· Use approved positive flow mask if significant quantities of dust becomes airborne. Try to avoid creating dust conditions.

Class P2 particulate filters are used for protection against mechanically and thermally generated particulates or both.

P2 is a respiratory filter rating under various international standards, Filters at least 94% of airborne particles Suitable for:

Continued...

 $\cdot$  Relatively small particles generated by mechanical processes eg. grinding, cutting, sanding, drilling, sawing.

 $\cdot$  Sub-micron thermally generated particles e.g. welding fumes, fertilizer and bushfire smoke.

 Biologically active airborne particles under specified infection control applications e.g. viruses, bacteria, COVID-19, SARS

# **SECTION 9 Physical and chemical properties**

## Information on basic physical and chemical properties

Appearance Organic brown liquid with lignin with acidic odour; mixes with water.

Physical state	Liquid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	4-5	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

# **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

# **SECTION 11 Toxicological information**

## Information on toxicological effects

Inhaled	Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful.
Ingestion	Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.
Skin Contact	Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. Exposure to copper, by skin, has come from its use in pigments, ointments, ornaments, jewellery, dental amalgams and IUDs (intra-uterine devices), and in killing fungi and algae. Although copper is used in the treatment of water in swimming pools and reservoirs, there are no reports of toxicity from these applications. This material can cause inflammation of the skin on contact in some persons.
Eye	If applied to the eyes, this material causes severe eye damage. Copper salts, in contact with the eye, may produce inflammation of the conjunctiva, or even ulceration and cloudiness of the cornea.
Chronic	Repeated or prolonged exposure to corrosives may result in the erosion of teeth, inflammatory and ulcerative changes in the mouth and necrosis (rarely) of the jaw. Bronchial irritation, with cough, and frequent attacks of bronchial pneumonia may ensue. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population. Toxic: danger of serious damage to health by prolonged exposure through inhalation, in contact with skin and if swallowed. This material can cause serious damage if one is exposed to it for long periods. It can be assumed that it contains a substance which can produce severe defects. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity results available. Animal testing shows that skin in exposure to copper may lead to hardness of the skin, scar formation, exudation and reddish changes. Inflammation, irritation and injury of the skin were noted. Repeat dose toxicity: Copper monochloride does not appear to cause mutations in vivo, although chromosomal aberrations were seen at very high concentrations in vitro. Cancer-causing potential: There was insufficient information to evaluate the cancer-causing activity of copper monochloride. Manganese is an essential trace element. Chronic exposure to low levels of manganese can include a mask-like facial expression, spastic gait, tremors, slurred speech, disordered muscle tone, fatigue, anorexia, loss of strength and energy, apathy and poor concentration. Welding or flame cutting of metals with zinc or zinc dust coatings may result in inhalation of zinc oxide fume; high concentrations of zinc oxide fume may result in "metal fume fever"; also known as "brass chills", an industrial disease of short duration. [I.L.O] Symptoms include

	susceptible people.	
	ΤΟΧΙΟΙΤΥ	IRRITATION
Smart Sulphate ZMC plus Moly	Not Available	Not Available
manganese sulfate, hydrate	TOXICITY Oral (Rat) LD50: 2150 mg/kg <sup>[2]</sup>	IRRITATION Not Available
	Orai (Rat) LDSU. 2 150 mg/kg 2	Not Available
zinc sulfate heptahydrate	ΤΟΧΙΟΙΤΥ	IRRITATION
	Oral (Mouse) LD50; 200 mg/kg <sup>[2]</sup>	Not Available
	ΤΟΧΙCΙΤΥ	IRRITATION
opper sulfate, pentahydrate	dermal (rat) LD50: >2000 mg/kg <sup>[2]</sup>	Not Available
	Oral (Mouse) LD50; 43 mg/kg <sup>[2]</sup>	
	τοχιςιτγ	IRRITATION
	dermal (rat) LD50: >2000 mg/kg <sup>[1]</sup>	Not Available
sodium molybdate	Inhalation (Rat) LC50: >1.93 mg/l4h <sup>[1]</sup>	
	Oral (Dog) LD50; 250 mg/kg <sup>[2]</sup>	
	ΤΟΧΙΟΙΤΥ	IRRITATION
	Dermal (rabbit) LD50: >1260 mg/kg <sup>[2]</sup>	Eye (rabbit): 119 mg - SEVERE [Monsanto]*
phosphoric acid	Inhalation (Rat) LC50: 0.026 mg/L4h <sup>[2]</sup>	Eye: adverse effect observed (irritating) <sup>[1]</sup>
	Oral (Rat) LD50: 1530 mg/kg <sup>[2]</sup>	Skin (rabbit):595 mg/24h - SEVERE
		Skin: adverse effect observed (corrosive) <sup>[1]</sup>
	RTECS criteria. for zinc sulfate heptahydrate Sleep, atax	
ZINC SULFATE HEPTAHYDRATE	pancreas recorded. Exposure may produce irreversible effects*. NOTE: Substance has been shown to be mutagenic in a to cellular DNA.	ta, respiratory stimulation, somnolence, coma, diarrhoea, changes in endocrine It least one assay, or belongs to a family of chemicals producing damage or chang

	causes excessive absorption and storage of copper. effects. Although it is excreted in the faeces, there is		
PHOSPHORIC ACID	phosphoric acid (85%) No significant acute toxicolog For acid mists, aerosols, vapours Test results suggest that eukaryotic cells are suscept have not been examined in this respect. Mucous sec (which also protects the stomach lining from the hydr The material may produce severe irritation to the eye produce conjunctivitis. The material may cause severe skin irritation after pr production of vesicles, scaling and thickening of the s	tible to genetic damage when the ph retion may protect the cells of the ai rochloric acid secreted there). a causing pronounced inflammation. rolonged or repeated exposure and i	I falls to about 6.5. Cells from the respiratory tract irway from direct exposure to inhaled acidic mists Repeated or prolonged exposure to irritants may may produce on contact skin redness, swelling, the
SODIUM MOLYBDATE & PHOSPHORIC ACID	Asthma-like symptoms may continue for months or e condition known as reactive airways dysfunction sym compound. Main criteria for diagnosing RADS includ of persistent asthma-like symptoms within minutes to include a reversible airflow pattern on lung function to and the lack of minimal lymphocytic inflammation, wi disorder with rates related to the concentration of and is a disorder that occurs as a result of exposure due reversible after exposure ceases. The disorder is character	drome (RADS) which can occur after e the absence of previous airways of b hours of a documented exposure tr ests, moderate to severe bronchial H thout eosinophilia. RADS (or asthma d duration of exposure to the irritatin to high concentrations of irritating su	er exposure to high levels of highly irritating disease in a non-atopic individual, with sudden onset o the irritant. Other criteria for diagnosis of RADS hyperreactivity on methacholine challenge testing, a) following an irritating inhalation is an infrequent g substance. On the other hand, industrial bronchitis ubstance (often particles) and is completely
Acute Toxicity	×	Carcinogenicity	×
Skin Irritation/Corrosion	✓	Reproductivity	×
Serious Eye	<b>~</b>		
Damage/Irritation	•	STOT - Single Exposure	×
	×	STOT - Single Exposure	× *

Data available to make classification

#### **SECTION 12 Ecological information**

Toxicity Endpoint Test Duration (hr) Species Value Source Smart Sulphate ZMC plus Not Not Not Moly Not Available Not Available Available Available Available Endpoint Test Duration (hr) Species Source Value LC50 Fish 130.465mg/l 4 96h 84mg/L 5 NOEC(ECx) 96h Fish 7.09-EC50 48h Crustacea 4 9.36mg/l manganese sulfate, hydrate EC50 96h Algae or other aquatic plants 25.7mg/l 4 EC50 72h Algae or other aquatic plants 61mg/l 2 2 NOEC(ECx) 1440h Crustacea 0.01mg/l 0.19-LC50 96h Fish 4 12.49mg/l Value Endpoint Test Duration (hr) Species Source Crustacea 0.04mg/L EC50 48h 5 zinc sulfate heptahydrate EC50(ECx) 120h Fish <0.001mg/L 5 LC50 96h Fish 0.103mg/l 4 Endpoint Test Duration (hr) Species Value Source EC50 48h Crustacea 0.003mg/L 5 EC50 72h Algae or other aquatic plants 0.8mg/L 5 copper sulfate, pentahydrate Crustacea EC50(ECx) 96h 0.001mg/L 5 LC50 96h Fish 0.073mg/L 4 Endpoint Test Duration (hr) Species Value Source 34.13-EC50 48h 4 Crustacea 46.87mg/l sodium molybdate EC50 2 72h Algae or other aquatic plants 26mg/l 2 NOEC(ECx) 672h Crustacea 0.67mg/l LC50 96h Fish >79.8mg/l 4 Endpoint Test Duration (hr) Species Value Source EC50 48h Crustacea >100mg/l 2 EC50 72h Algae or other aquatic plants 77.9mg/l 2 phosphoric acid 2 NOEC(ECx) 72h Algae or other aquatic plants <7.5mg/l 67.94-LC50 96h 4 Fish 113.76mg/L

Legend: Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

For Copper Sulfate:

Terrestrial Fate: Soil - If released to soil, copper sulfate may leach to groundwater and may partially oxidize or bind to humic materials, clay or hydrous oxides of iron and manganese. Since copper is an element, it will persist indefinitely. Copper is bound or adsorbed, to organic materials, and to clay and mineral surfaces. The degree of adsorption to soils depends on the acidity or alkalinity of the soil. Copper sulfate is one of the more mobile metals in soil; however, its leaching potential is low in all but sandy soils. When applied with irrigation water, copper sulfate does not accumulate in the surrounding soils; however, some 60% is deposited in the sediments at the bottom of the irrigation ditch, where it becomes adsorbed to clay, mineral, and organic particles. Copper compounds also settle out of solution. Plants – Copper sulfate is toxic to plants and kills by photosynthesis disruption. Blue-green algae have been shown to become increasingly resistant to the algaecide after 26 years of use.

Aquatic Fate: In water, copper sulfate will bind to carbonates as well as humic materials, clay and hydrous oxides of iron and manganese. As an element, copper can persist indefinitely.

Ecotoxicity: Copper is accumulated by plants and animals, but, it does not appear to biomagnify from plants to animals. Copper sulphate is practically non-toxic to birds and poses less of a threat to birds than to other animals. Copper sulfate is highly toxic to fish and Daphnia magna water fleas. Even at recommended rates of application, this material may be poisonous to trout and other fish, especially in soft or acid waters. Its toxicity to fish generally decreases as water hardness increases. Fish eggs are more resistant than young fish fry to the toxic effects of copper sulfate. Copper sulfate is toxic to aquatic invertebrates, such as crab, shrimp, and oysters. Higher concentrations of the material caused some behavioral changes, such as secretion of mucous, and discharge of eggs and embryos. Bees are endangered by Bordeaux mixtures of copper (II) sulfate and hydrated lime. Copper sulfate may be poisonous to sheep and chickens at normal application rates. Most animal life in soil, including large earthworms, have been eliminated by the extensive use of copper containing fungicides in orchards. **DO NOT** discharge into sewer or waterways.

#### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
zinc sulfate heptahydrate	HIGH	HIGH
copper sulfate, pentahydrate	HIGH	HIGH
sodium molybdate	HIGH	HIGH
phosphoric acid	HIGH	HIGH

#### **Bioaccumulative potential**

Ingredient	Bioaccumulation
zinc sulfate heptahydrate	LOW (BCF = 112)
copper sulfate, pentahydrate	LOW (LogKOW = -2.2002)
sodium molybdate	LOW (LogKOW = 2.229)
phosphoric acid	LOW (LogKOW = -0.7699)

#### Mobility in soil

Ingredient	Mobility
zinc sulfate heptahydrate	LOW (Log KOC = 6.124)
copper sulfate, pentahydrate	LOW (Log KOC = 6.124)
sodium molybdate	LOW (Log KOC = 48.64)
phosphoric acid	HIGH (Log KOC = 1)

#### **SECTION 13 Disposal considerations**

Waste treatment methods	
Product / Packaging disposal	<ul> <li>Containers may still present a chemical hazard/ danger when empty.</li> <li>Return to supplier for reuse/ recycling if possible.</li> <li>Otherwise: <ul> <li>If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.</li> <li>Where possible retain label warnings and SDS and observe all notices pertaining to the product.</li> <li>DO NOT allow wash water from cleaning or process equipment to enter drains.</li> <li>It may be necessary to collect all wash water for treatment before disposal.</li> <li>In all cases disposal to sever may be subject to local laws and regulations and these should be considered first.</li> <li>Where in doubt contact the responsible authority.</li> <li>Recycle wherever possible or consult manufacturer for recycling options.</li> <li>Consult State Land Waste Authority for disposal.</li> <li>Bury or incinerate residue at an approved site.</li> <li>Recycle containers if possible, or dispose of in an authorised landfill.</li> </ul> </li> </ul>

#### **SECTION 14 Transport information**

#### Labels Required



Marine Pollutant	
HAZCHEM	•3Z
Land transport (ADG)	
14.1. UN number or ID number	3082
14.2. UN proper shipping	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)

name		
14.3. Transport hazard class(es)	Class9Subsidiary HazardNot Applicable	
14.4. Packing group	II	
14.5. Environmental hazard	Environmentally hazardous	
14.6. Special precautions for user	Special provisions         274 331 335 375 AU01           Limited quantity         5 L	

Environmentally Hazardous Substances meeting the descriptions of UN 3077 or UN 3082 are not subject to this Code when transported by road or rail in;

(a) packagings;
(b) IBCs; or
(c) any other receptacle not exceeding 500 kg(L).
- Australian Special Provisions (SP AU01) - ADG Code 7th Ed.

## Air transport (ICAO-IATA / DGR)

14.1. UN number	3082			
14.2. UN proper shipping name	Environmentally hazardous substance, liquid, n.o.s. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)			
	ICAO/IATA Class	9		
14.3. Transport hazard class(es)	ICAO / IATA Subsidiary Hazard	Not Applicable		
class(es)	ERG Code	9L		
14.4. Packing group	Ш			
14.5. Environmental hazard	Environmentally hazardous			
14.6. Special precautions for user	Special provisions		A97 A158 A197 A215	
	Cargo Only Packing Instructions		964	
	Cargo Only Maximum Qty / Pack		450 L	
	Passenger and Cargo Packing Instructions		964	
	Passenger and Cargo Maximum Qty / Pack		450 L	
	Passenger and Cargo Limited Qu	antity Packing Instructions	Y964	
	Passenger and Cargo Limited Maximum Qty / Pack		30 kg G	

## Sea transport (IMDG-Code / GGVSee)

14.1. UN number	3082		
14.2. UN proper shipping name	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)		
14.3. Transport hazard class(es)	IMDG Class IMDG Subsidiary Ha	9 azard Not Applicable	
14.4. Packing group			
14.5 Environmental hazard	Marine Pollutant		
14.6. Special precautions for user	EMS Number	F-A, S-F	
	Special provisions	274 335 969	
	Limited Quantities	5 L	

## 14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

# 14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
manganese sulfate, hydrate	Not Available
zinc sulfate heptahydrate	Not Available
copper sulfate, pentahydrate	Not Available
sodium molybdate	Not Available
phosphoric acid	Not Available

#### 14.7.3. Transport in bulk in accordance with the IGC Code

Ship Type
Not Available

## **SECTION 15 Regulatory information**

## Safety, health and environmental regulations / legislation specific for the substance or mixture

# manganese sulfate, hydrate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australian Inventory of Industrial Chemicals (AIIC)

zinc sulfate heptahydrate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 6 Australian Inventory of Industrial Chemicals (AIIC)

#### copper sulfate, pentahydrate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 5 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 6 Australia Inventory of Industrial Chemicals (AIIC)

#### sodium molybdate is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

#### phosphoric acid is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 5 Australian Inventory of Industrial Chemicals (AIIC)

#### Additional Regulatory Information

Not Applicable

#### **National Inventory Status**

National Inventory	Status	
Australia - AIIC / Australia Non- Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (manganese sulfate, hydrate; zinc sulfate heptahydrate; copper sulfate, pentahydrate; sodium molybdate; phosphoric acid)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

#### **SECTION 16 Other information**

Revision Date	28/03/2024
Initial Date	28/03/2024

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
- TEEL: Temporary Emergency Exposure Limit,
   IDLH: Immediately Dangerous to Life or Health Concentrations
- ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level
- LOAEL: Lowest Observed Adverse Effect Level
- TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value
- BCF: BioConcentration Factors BEI: Biological Exposure Index
- DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration
- AlIC: Australian Inventory of Industrial Chemicals
- DSL: Domestic Substances List
- NDSL: Non-Domestic Substances List
- IECSC: Inventory of Existing Chemical Substance in China
   EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
- ENCS: Existing and New Chemical Substances Inventory
- KECI: Korea Existing Chemicals Inventory
   NZIOC: New Zealand Inventory of Chemicals
- PICCS: Philippine Inventory of Chemicals and Chemical Substances
- TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory
- INSQ: Inventario Nacional de Sustancias Químicas
- NCI: National Chemical Inventory
   FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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