Crop Smart better crop protection

Smart Select Carbo King

Crop Smart Pty Ltd

Chemwatch: 5663-37 Version No: 2.1 Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements

Issue Date: 14/03/2024 Print Date: 26/03/2024 S.GHS.AUS.EN.E

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier	
Product name	Smart Select Carbo King
Chemical Name	Not Applicable
Synonyms	Not Available
Proper shipping name	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)
Chemical formula	Not Applicable
Other means of identification	Not Available

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Foliar Fertiliser for application in agriculture.
Relevant identified uses	Use according to manufacturer's directions.

Details of the manufacturer or supplier of the safety data sheet

Registered company name	Crop Smart Pty Ltd	
Address	2409/ 4 Daydream Street WARRIEWOOD NSW 2102 Australia	
Telephone	+61 1300 783 481	
Fax	Not Available	
Website	www.cropsmart.com.au	
Email	Compliance@cropsmart.com.au	

Emergency telephone number

Association / Organisation	CHEMWATCH EMERGENCY RESPONSE (24/7)	
Emergency telephone numbers	+61 1800 951 288	
Other emergency telephone numbers	+61 3 9573 3188	

Once connected and if the message is not in your preferred language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture

Poisons Schedule	S6
Classification ^[1]	Acute Toxicity (Oral) Category 4, Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 1, Acute Toxicity (Inhalation) Category 4, Specific Target Organ Toxicity - Repeated Exposure Category 2, Hazardous to the Aquatic Environment Long-Term Hazard Category 1
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

Label elements

Hazard pictogram(s)

Signal word

Danger

Hazard statement(s)

H302	Harmful if swallowed.
H315	Causes skin irritation.
H318	Causes serious eye damage.
H332	Harmful if inhaled.

H373 May cause damage to organs through prolonged or repeated exposure.		
H410	Very toxic to aquatic life with long lasting effects.	
Precautionary statement(s) Prevention		

P260	P260 Do not breathe mist/vapours/spray.	
P271	Use only outdoors or in a well-ventilated area.	
P280	280 Wear protective gloves, protective clothing, eye protection and face protection.	
P264	P264 Wash all exposed external body areas thoroughly after handling.	
P270	Do not eat, drink or smoke when using this product.	
P273	Avoid release to the environment.	

Precautionary statement(s) Response

P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.		
P310	Immediately call a POISON CENTER/doctor/physician/first aider.		
P391	Collect spillage.		
P301+P312	IF SWALLOWED: Call a POISON CENTER/doctor/physician/first aider if you feel unwell.		
P302+P352	IF ON SKIN: Wash with plenty of water.		
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.		
P330	Rinse mouth.		
P332+P313	If skin irritation occurs: Get medical advice/attention.		
P362+P364	Take off contaminated clothing and wash it before reuse.		

Precautionary statement(s) Storage

Not Applicable

Precautionary statement(s) Disposal

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

P501

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
10034-96-5	10-30	manganese sulfate, hydrate
7446-20-0	10-30	zinc sulfate heptahydrate
7664-38-2	10-30	phosphoric acid
7758-99-8	1-10	copper sulfate, pentahydrate
7782-63-0	1-5	ferrous sulfate heptahydrate
57-13-6	<5	urea
10102-40-6	<1	sodium molybdate
Not Available	balance	Ingredients determined not to be hazardous
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available	

SECTION 4 First aid measures

Description of first aid measure	es
Eye Contact	 If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor, without delay.
Ingestion	 For advice, contact a Poisons Information Centre or a doctor at once. Urgent hospital treatment is likely to be needed. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully.

- Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.
- F Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.
- Transport to hospital or doctor without delay.

Indication of any immediate medical attention and special treatment needed

for phosphate salts intoxication:

- All treatments should be based on observed signs and symptoms of distress in the patient. Consideration should be given to the possibility that overexposure to materials other than this product may have occurred.
- Ingestion of large quantities of phosphate salts (over 1.0 grams for an adult) may cause an osmotic catharsis resulting in diarrhoea and probable abdominal cramps. Larger doses such as 4-8 grams will almost certainly cause these effects in everyone. In healthy individuals most of the ingested salt will be excreted in the faeces with the diarrhoea and, thus, not cause any systemic toxicity. Doses greater than 10 grams hypothetically may cause systemic toxicity.
- Treatment should take into consideration both anionic and cation portion of the molecule.
- All phosphate salts, except calcium salts, have a hypothetical risk of hypocalcaemia, so calcium levels should be monitored.

Treat symptomatically. for copper intoxication:

- Unless extensive vomiting has occurred empty the stomach by lavage with water, milk, sodium bicarbonate solution or a 0.1% solution of potassium ferrocyanide (the resulting copper ferrocyanide is insoluble).
- Administer egg white and other demulcents.
- Maintain electrolyte and fluid balances.
- Morphine or meperidine (Demerol) may be necessary for control of pain.
- If symptoms persist or intensify (especially circulatory collapse or cerebral disturbances, try BAL intramuscularly or penicillamine in accordance with the supplier's recommendations.
- Treat shock vigorously with blood transfusions and perhaps vasopressor amines.
- If intravascular haemolysis becomes evident protect the kidneys by maintaining a diuresis with mannitol and perhaps by alkalinising the urine with sodium bicarbonate.
- It is unlikely that methylene blue would be effective against the occassional methaemoglobinemia and it might exacerbate the subsequent haemolytic episode.
- Institute measures for impending renal and hepatic failure.
- [GOSSELIN, SMITH & HODGE: Commercial Toxicology of Commercial Products]
- A role for activated charcoals for emesis is, as yet, unproven.
- In severe poisoning CaNa2EDTA has been proposed.
- [ELLENHORN & BARCELOUX: Medical Toxicology]

For acute or short term repeated exposures to iron and its derivatives:

- Always treat symptoms rather than history.
- In general, however, toxic doses exceed 20 mg/kg of ingested material (as elemental iron) with lethal doses exceeding 180 mg/kg.
- Control of iron stores depend on variation in absorption rather than excretion. Absorption occurs through aspiration, ingestion and burned skin.
- Hepatic damage may progress to failure with hypoprothrombinaemia and hypoglycaemia. Hepatorenal syndrome may occur.
- Iron intoxication may also result in decreased cardiac output and increased cardiac pooling which subsequently produces hypotension.
- Serum iron should be analysed in symptomatic patients. Serum iron levels (2-4 hrs post-ingestion) greater that 100 ug/dL indicate poisoning with levels, in excess of 350 ug/dL, being potentially serious. Emesis or lavage (for obtunded patients with no gag reflex) are the usual means of decontamination.
- Activated charcoal does not effectively bind iron.
- Catharsis (using sodium sulfate or magnesium sulfate) may only be used if the patient already has diarrhoea.
- Deferoxamine is a specific chelator of ferric (3+) iron and is currently the antidote of choice. It should be administered parenterally. [Ellenhorn and Barceloux: Medical Toxicology]

Both dermal and oral toxicity of manganese salts is low because of limited solubility of manganese. No known permanent pulmonary sequelae develop after acute manganese exposure. Treatment is supportive.

[Ellenhorn and Barceloux: Medical Toxicology]

In clinical trials with miners exposed to manganese-containing dusts, L-dopa relieved extrapyramidal symptoms of both hypo kinetic and dystonic patients. For short periods of time symptoms could also be controlled with scopolamine and amphetamine. BAL and calcium EDTA prove ineffective.

[Gosselin et al: Clinical Toxicology of Commercial Products.]

- For acute or short term repeated exposures to strong acids:
- Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling
- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise
- Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues.

INGESTION:

- Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.
- DO NOT attempt to neutralise the acid since exothermic reaction may extend the corrosive injury.
- Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
- Charcoal has no place in acid management.
- Some authors suggest the use of lavage within 1 hour of ingestion.
- SKIN:
- Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.
- Deep second-degree burns may benefit from topical silver sulfadiazine.

EYE:

- Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjuctival cul-de-sacs. Irrigation should last at least 20-30 minutes. DO NOT use neutralising agents or any other additives. Several litres of saline are required.
- Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury.
- Steroid eye drops should only be administered with the approval of a consulting ophthalmologist)

[Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 Firefighting measures

Extinguishing media

- Foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.
 Water spray or fog Large fires only.

Special hazards arising from the substrate or mixture

Fire Incompatibility + Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

Advice for firefighters

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Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Use water delivered as a fine spray to control fire and cool adjacent area. Avoid spraying water onto liquid pools. DO NOT approach containers with water spray from a protected location. If safe to do so, remove containers from path of fire.
Fire/Explosion Hazard	 Combustible. Slight fire hazard when exposed to heat or flame. Acids may react with metals to produce hydrogen, a highly flammable and explosive gas. Heating may cause expansion or decomposition leading to violent rupture of containers. May emit acrid smoke and corrosive fumes. Combustion products include: carbon dioxide (CO2) phosphorus oxides (POx) sulfur oxides (SO2) metal oxides other pyrolysis products typical of burning organic material.
HAZCHEM	•3Z

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Environmental hazard - contain spillage. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb spill with sand, earth, inert material or vermiculite. Wipe up. Place in a suitable, labelled container for waste disposal.
Major Spills	 Environmental hazard - contain spillage. Moderate hazard. Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water course. No smoking, naked lights or ignition sources. Increase ventilation. Stop leak if safe to do so. Contain spill with sand, earth or vermiculite. Collect recoverable product into labelled containers for recycling. Absorb remaining product with sand, earth or vermiculite. Collect solid residues and seal in labelled drums for disposal. Wash area and prevent runoff into drains. If contamination of drains or waterways occurs, advise emergency services.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Prevent concentration in hollows and sumps. DO NOT enter confined spaces until atmosphere has been checked. Avoid smoking, naked lights or ignition sources. Avoid contact with incompatible materials. When handling, DO NOT eat, drink or smoke. Keep containers securely sealed when not in use. Avoid physical damage to containers. Always wash hands with soap and water after handling. Work clothes should be laundered separately. Use good occupational work practice. Observe manufacturer's storage and handling recommendations contained within this SDS. Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions.
Other information	 Store in original containers. Keep containers securely sealed. No smoking, naked lights or ignition sources. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this SDS.

Suitable container	 Glass container is suitable for laboratory quantities DO NOT use aluminium or galvanised containers Lined metal can, lined metal pail/ can. Plastic pail. Polyliner drum. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks. Metal can or drum Packaging as recommended by manufacturer. Check all containers are clearly labelled and free from leaks. Metal containers are clearly labelled and free from leaks.
Storage incompatibility	Avoid reaction with oxidising agents

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	manganese sulfate, hydrate	Manganese, dust & compounds (as Mn)	1 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	phosphoric acid	Phosphoric acid	1 mg/m3	3 mg/m3	Not Available	Not Available
Australia Exposure Standards	ferrous sulfate heptahydrate	Iron salts, soluble (as Fe)	1 mg/m3	Not Available	Not Available	Not Available
Australia Exposure Standards	sodium molybdate	Molybdenum, soluble compounds (as Mo)	5 mg/m3	Not Available	Not Available	Not Available

	I.	Emergency	Limits
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Ingredient	TEEL-1	TEEL-2		TEEL-3	
manganese sulfate, hydrate	9.2 mg/m3	15 mg/m3		90 mg/m3	
manganese sulfate, hydrate	8.2 mg/m3	14 mg/m3		430 mg/m3	
zinc sulfate heptahydrate	27 mg/m3	170 mg/m3		1,000 mg/m3	
zinc sulfate heptahydrate	15 mg/m3	97 mg/m3		580 mg/m3	
phosphoric acid	Not Available	Not Available		Not Available	
copper sulfate, pentahydrate	7.5 mg/m3	9.9 mg/m3		59 mg/m3	
copper sulfate, pentahydrate	12 mg/m3	32 mg/m3		190 mg/m3	
ferrous sulfate heptahydrate	8.2 mg/m3	41 mg/m3		250 mg/m3	
ferrous sulfate heptahydrate	15 mg/m3	170 mg/m3		990 mg/m3	
urea	30 mg/m3	280 mg/m3		1,700 mg/m3	
sodium molybdate	3.8 mg/m3 34 mg/m3			210 mg/m3	
sodium molybdate	3.2 mg/m3	17 mg/m3		100 mg/m3	
Ingredient	Original IDLH		Revised IDLH		
manganese sulfate, hydrate	500 mg/m3		Not Available		
zinc sulfate heptahydrate	Not Available		Not Available		
phosphoric acid	1,000 mg/m3		Not Available		
copper sulfate, pentahydrate	Not Available		Not Available		
ferrous sulfate heptahydrate	Not Available		Not Available		
urea	Not Available	Not Available		Not Available	
sodium molybdate	1,000 mg/m3		Not Available		

Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating Occupational Exposure Band Limit		
zinc sulfate heptahydrate	≤ 0.01 mg/m³		
copper sulfate, pentahydrate	≤ 0.01 mg/m³		
urea	E ≤ 0.01 mg/m³		
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the		

Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a range of exposure concentrations that are expected to protect worker health.

Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the ha can be highly effective in protecting workers and will typically be independent of worker interactions. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away fror strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an design of a ventilation system must match the particular process and chemical or contaminant in us Employers may need to use multiple types of controls to prevent employee overexposure. Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator, or protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essee An approved self contained breathing apparatus (SCBA) may be required in some situations. Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectiv	to provide this high level of protection.
	Type of Contaminant:	Air Speed:

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	solvent, vapours, degreasing etc., evaporating from tank (ir	n still air).	0.25-0.5 m/s (50- 100 f/min.)		
	aerosols, fumes from pouring operations, intermittent conta spray drift, plating acid fumes, pickling (released at low velo		0.5-1 m/s (100- 200 f/min.)		
	direct spray, spray painting in shallow booths, drum filling, o generation into zone of rapid air motion)	1-2.5 m/s (200- 500 f/min.)			
	grinding, abrasive blasting, tumbling, high speed wheel gen of very high rapid air motion).	2.5-10 m/s (500- 2000 f/min.)			
	Within each range the appropriate value depends on:				
	Lower end of the range	Upper end of the range			
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents			
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity			
	3: Intermittent, low production.	3: High production, heavy use			
	4: Large hood or large air mass in motion	4: Small hood-local control only			
	Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity gener decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for exampl a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. If mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air v multiplied by factors of 10 or more when extraction systems are installed or used.				
Individual protection measures, such as personal protective equipment					
Eye and face protection	 are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under pressure. Chemical goggles. Whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted. [AS/NZS 1337.1, EN166 or national equivalent] Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection. Alternatively a gas mask may replace splash goggles and face shields. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59]. See Hand protection below Elbow length PVC gloves 				
Skin protection	See Hand protection below				
Hands/feet protection	 NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed. The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dired thoroughly. Application of a non-perfumed moisturiser is recommended. Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include: frequency and duration of contact, chemical resistance of glove material, glove thickness and dexterity Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent). When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended. Some glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term use. contaminated gloves should be replaced. As defined in ASTM F.739-66 in any application, gloves are rated as: Excellent when breakthrough time < 20 min Fair when brea				

Other protection	 Overalls. P.V.C apron. Barrier cream. Skin cleansing cream. Eye wash unit.
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Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the computergenerated selection:

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Material	СРІ
NAT+NEOPR+NITRILE	А
NATURAL RUBBER	А
NATURAL+NEOPRENE	А
NEOPRENE	А
NEOPRENE/NATURAL	А
NITRILE	А
NITRILE+PVC	А
PE	А
PVC	А
SARANEX-23	A

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion **NOTE**: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted

Ansell Glove Selection

Glove — In order of recommendation
AlphaTec® Solvex® 37-675
AlphaTec 02-100
AlphaTec® Solvex® 37-185
AlphaTec® 58-008
AlphaTec® 58-530B
AlphaTec® 58-530W
AlphaTec® 58-735
AlphaTec® 79-700
AlphaTec® 38-612
MICROFLEX® 63-864

The suggested gloves for use should be confirmed with the glove supplier.

Respiratory protection

Type AB-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	AB-AUS P2	-	AB-PAPR-AUS / Class 1 P2
up to 50 x ES	-	AB-AUS / Class 1 P2	-
up to 100 x ES	-	AB-2 P2	AB-PAPR-2 P2 ^

^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%. in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

· The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option). · Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.

· Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.

· Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved

under appropriate government standards such as NIOSH (US) or CEN (EU) · Use approved positive flow mask if significant quantities of dust becomes airborne.

Try to avoid creating dust conditions Class P2 particulate filters are used for protection against mechanically and thermally generated particulates or both.

P2 is a respiratory filter rating under various international standards, Filters at least 94% of airborne particles

Suitable for:

· Relatively small particles generated by mechanical processes eg. grinding, cutting, sanding, drilling, sawing.

· Sub-micron thermally generated particles e.g. welding fumes, fertilizer and bushfire smoke

Biologically active airborne particles under specified infection control applications e.g. viruses, bacteria, COVID-19, SARS

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SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Organic brown liquid with lignin with acidic like odour; mixes with water.			
Physical state	Liquid	Relative density (Water = 1)	Not Available	
Odour	Not Available	Partition coefficient n-octanol / water	Not Available	
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available	
pH (as supplied)	2-3	Decomposition temperature (°C)	Not Available	
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available	

Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

phosphoric acid

Information on toxicological effects

· · · · · · · · · · · · · · · · · · ·			
Inhaled	Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful. The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.		
Ingestion	Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.		
Skin Contact	Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. The material may cause severe inflammation of the skin either following direct contact or after a delay of some time. Repeated exposure can cause contact dermatitis which is characterised by redness, swelling and blistering.		
Eye	If applied to the eyes, this material causes severe eye damage.		
Chronic	lung, with cough, and inflammation of lung tissue often occurs. For copper and its compounds (typically copper chloride): Acute toxicity: There are no reliable acute oral toxicity results availabl hardness of the skin, scar formation, exudation and reddish changes. Repeat dose toxicity: Animal testing shows that very high levels of co Genetic toxicity: Copper monochloride does not appear to cause mut high concentrations in vitro. Cancer-causing potential: There was insufficient information to evalue Manganese is an essential trace element. Chronic exposure to low le gait, tremors, slurred speech, disordered muscle tone, fatigue, anore: Chronic excessive intake of iron have been associated with damage control over iron are at an increased risk. Welding or flame cutting of metals with zinc or zinc dust coatings may	eaction in some persons compared to the general population. ough inhalation, in contact with skin and if swallowed. ag periods. It can be assumed that it contains a substance which can if teeth, swelling and/or ulceration of mouth lining. Irritation of airways to e. Animal testing shows that skin in exposure to copper may lead to Inflammation, irritation and injury of the skin were noted. pper monochloride may cause anaemia. ations in vivo, although chromosomal aberrations were seen at very ate the cancer-causing activity of copper monochloride. vels of manganese can include a mask-like facial expression, spastic kia, loss of strength and energy, apathy and poor concentration. to the liver and pancreas. People with a genetic disposition to poor <i>r</i> result in inhalation of zinc oxide fume; high concentrations of zinc ills", an industrial disease of short duration. [I.L.O] Symptoms include ns occur in enclosed or poorly ventilated areas. nulative health effects involving organs or biochemical systems.	
	ΤΟΧΙΟΙΤΥ	IRRITATION	
Smart Select Carbo King	Not Available	Not Available	
manganese sulfate, hydrate		IRRITATION	
	Oral (Rat) LD50: 2150 mg/kg ^[2]	Not Available	
Tino cultoto hontohudroto	ΤΟΧΙΟΙΤΥ	IRRITATION	
zinc sulfate heptahydrate	$Oral (Mouse) \mid D50; 200 mg/kg^{[2]}$	Not Available	

Oral (Mouse) LD50; 200 mg/kg^[2]

Dermal (rabbit) LD50: >1260 mg/kg^[2] Inhalation (Rat) LC50: 0.026 mg/L4h^[2]

Oral (Rat) LD50: 1530 mg/kg^[2]

ΤΟΧΙΟΙΤΥ

Not Available

IRRITATION

		Skin: adverse effect observed (corrosive) $\ensuremath{\left[1\right]}$
	ΤΟΧΙΟΙΤΥ	IRRITATION
opper sulfate, pentahydrate	dermal (rat) LD50: >2000 mg/kg ^[2]	Not Available
	Oral (Mouse) LD50; 43 mg/kg ^[2]	
	ΤΟΧΙΟΙΤΥ	IRRITATION
errous sulfate heptahydrate	Oral (Mouse) LD50; 1520 mg/kg ^[2]	Not Available
	ΤΟΧΙΟΙΤΥ	IRRITATION
urea	dermal (rat) LD50: 8200 mg/kg ^[2]	Eye: no adverse effect observed (not irritating) ^[1]
	Oral (Rat) LD50: 8471 mg/kg ^[2]	Skin (human): 22 mg/3 d (I)- mild
		Skin: no adverse effect observed (not irritating) ^[1]
	ΤΟΧΙΟΙΤΥ	IRRITATION
	dermal (rat) LD50: >2000 mg/kg ^[1]	Not Available
sodium molybdate	Inhalation (Rat) LC50: >1.93 mg/l4h ^[1]	
	Oral (Dog) LD50; 250 mg/kg ^[2]	
Legend:	1. Value obtained from Europe ECHA Registered Substances - A specified data extracted from RTECS - Register of Toxic Effect of	cute toxicity 2. Value obtained from manufacturer's SDS. Unless otherwis chemical Substances
MANGANESE SULFATE, HYDRATE	Not available.	
ZINC SULFATE HEPTAHYDRATE	RTECS criteria. for zinc sulfate heptahydrate Sleep, ataxia, respir pancreas recorded. Exposure may produce irreversible effects*.	ate Oral (man) TDLo: 180 mg/kg/6w-I Equivocal tumorigenic agent by atory stimulation, somnolence, coma, diarrhoea, changes in endocrine ne assay, or belongs to a family of chemicals producing damage or change
PHOSPHORIC ACID	have not been examined in this respect. Mucous secretion may p (which also protects the stomach lining from the hydrochloric acic The material may produce severe irritation to the eye causing pro produce conjunctivitis.	ic damage when the pH falls to about 6.5. Cells from the respiratory tract rotect the cells of the airway from direct exposure to inhaled acidic mists I secreted there). nounced inflammation. Repeated or prolonged exposure to irritants may epeated exposure and may produce on contact skin redness, swelling, the
COPPER SULFATE, PENTAHYDRATE	male rats and 5 groups of 5 female rats received doses of 1000, values of copper monochloride were 2,000 mg/kg bw or greater fn died at both 1500 and 2000 mg/kg bw, and one at 1,000 mg/kg brows and the second sec	CE frequencies compared to those of the negative control animals.

Copper sulfate is corrosive. Side effects are diverse and multi-systemic, and include severe gastrointestinal symptoms and signs, metallic taste in the mouth, burning pain in the chest, headache, sweating, shock and damage to brain, liver and kidneys. It has been reported as a cause of human suicide. On exposure, it can cause dose dependent damage to the skin and eye, also, eczema and allergic reactions. Long term effects can lead to anaemia and degenerative changes and are more likely in individuals with Wilson's disease, a condition which causes excessive absorption and storage of copper. It has adverse effects on reproduction and fertility as well as cancer and embryo toxic effects. Although it is excreted in the faeces, there is residual accumulation the liver, brain, heart, kidney and muscles.

Altered sleep time, change in motor activity, antipsychosis, dyspnea, methaemoglobinaemia, convulsions, lymphomas recorded. Carcinogenic by RTECS criteria.

	Based on laboratory and animal testing, exposure to The material may cause skin irritation after prolonge production of vesicles, scaling and thickening of the For urea: Urea is used in ointments and creams to treat dry sk allergy, and is virtually free from side effects. It is us amounts (60-90 grams/day). There is the possibility because of the generation of ammonia. Acute toxicity: Animal testing shows that the acute to Repeated dose toxicity: No well-conducted repeated toxicity. Reproductive and developmental toxicity: No adequa Genetic toxicity: Urea has been negative in several a mammals, it causes chromosomal aberrations only a	d or repeated exposure and may pro- skin. kin. Long-term follow-up studies have ually tolerated well, although diarrhea that infection of H. pylori in the huma exicity of urea is low. I dose toxicity studies were located. ate data exists regarding the reprodu appropriately conducted tests on bac	aduce on contact skin redness, swelling, the e indicated that the substance does not cause a is sometimes reported after ingestion of very large an stomach may aggravate local effects by urea Tests involving the skin on animals suggested low active/developmental toxicity of urea.
PHOSPHORIC ACID & UREA & SODIUM MOLYBDATE	Asthma-like symptoms may continue for months or e condition known as reactive airways dysfunction sym compound. Main criteria for diagnosing RADS includ of persistent asthma-like symptoms within minutes tr include a reversible airflow pattern on lung function f and the lack of minimal lymphocytic inflammation, w disorder with rates related to the concentration of an is a disorder that occurs as a result of exposure due reversible after exposure ceases. The disorder is ch	drome (RADS) which can occur after the absence of previous airways of o hours of a documented exposure to tests, moderate to severe bronchial h ithout eosinophilia. RADS (or asthma d duration of exposure to the irritatin to high concentrations of irritating su	er exposure to high levels of highly irritating lisease in a non-atopic individual, with sudden onset o the irritant. Other criteria for diagnosis of RADS hyperreactivity on methacholine challenge testing, a) following an irritating inhalation is an infrequent g substance. On the other hand, industrial bronchitis ubstance (often particles) and is completely
Acute Toxicity	~	Carcinogenicity	×
Skin Irritation/Corrosion	✓	Reproductivity	×
Serious Eye Damage/Irritation	×	STOT - Single Exposure	×
Respiratory or Skin	×	STOT - Repeated Exposure	*
sensitisation			

SECTION 12 Ecological information

	Endpoint	Test Duration (hr)	Species	Value	Source
Smart Select Carbo King	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
	LC50	96h	Fish	130.465mg/l	4
	NOEC(ECx)	96h	Fish	84mg/L	5
	EC50	48h	Crustacea	7.09- 9.36mg/l	4
manganese sulfate, hydrate	EC50	96h	Algae or other aquatic plants	25.7mg/l	4
	EC50	72h	Algae or other aquatic plants	61mg/l	2
	NOEC(ECx)	1440h	Crustacea	0.01mg/l	2
	LC50	96h	Fish	0.19- 12.49mg/l	4
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	EC50	48h	Crustacea	0.04mg/L	5
zinc sulfate heptahydrate	EC50(ECx)	120h	Fish	<0.001mg/L	5
	LC50	96h	Fish	0.103mg/l	4
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	EC50	48h	Crustacea	>100mg/l	2
phosphoric acid	EC50	72h	Algae or other aquatic plants	77.9mg/l	2
phosphone actu	NOEC(ECx)	72h	Algae or other aquatic plants	<7.5mg/l	2
	LC50	96h	Fish	67.94- 113.76mg/L	4
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	EC50	48h	Crustacea	0.003mg/L	5
copper sulfate, pentahydrate	EC50	72h	Algae or other aquatic plants	0.8mg/L	5
	EC50(ECx)	96h	Crustacea	0.001mg/L	5
	LC50	96h	Fish	0.073mg/L	4
ferrous sulfate heptahydrate	Endpoint	Test Duration (hr)	Species	Value	Sourc
	EC50	48h	Crustacea	12.35- 16.72mg/L	4
	EC50(ECx)	48h	Crustacea	12.35- 16.72mg/L	4

	LC50	96h	Fish	6.27- 50.35mg/L	4
	Endpoint	Test Duration (hr)	Species	Value	Source
	ErC50	72h	Algae or other aquatic plants	24541.9mg/l	2
	EC50	48h	Crustacea	3910mg/l	4
urea	EC50	72h	Algae or other aquatic plants	24541.9mg/l	2
	NOEC(ECx)	5040h	Fish	>=1.71mg/l	2
	LC50	96h	Fish	4.65- 8.48mg/l	4
	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	48h	Crustacea	34.13- 46.87mg/l	4
sodium molybdate	EC50	72h	Algae or other aquatic plants	26mg/l	2
	NOEC(ECx)	672h	Crustacea	0.67mg/l	2
	NOEC(ECx) LC50	672h 96h	Crustacea Fish	0.67mg/l >79.8mg/l	2

Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
zinc sulfate heptahydrate	HIGH	HIGH
phosphoric acid	HIGH	HIGH
copper sulfate, pentahydrate	HIGH	HIGH
ferrous sulfate heptahydrate	HIGH	HIGH
urea	LOW	LOW
sodium molybdate	HIGH	HIGH

Bioaccumulative potential

Ingredient	Bioaccumulation
zinc sulfate heptahydrate	LOW (BCF = 112)
phosphoric acid	LOW (LogKOW = -0.7699)
copper sulfate, pentahydrate	LOW (LogKOW = -2.2002)
ferrous sulfate heptahydrate	LOW (BCF = 52)
urea	LOW (BCF = 10)
sodium molybdate	LOW (LogKOW = 2.229)

Mobility in soil

Ingredient	Mobility
zinc sulfate heptahydrate	LOW (Log KOC = 6.124)
phosphoric acid	HIGH (Log KOC = 1)
copper sulfate, pentahydrate	LOW (Log KOC = 6.124)
ferrous sulfate heptahydrate	LOW (Log KOC = 6.124)
urea	LOW (Log KOC = 4.191)
sodium molybdate	LOW (Log KOC = 48.64)

SECTION 13 Disposal considerations

Waste treatment methods	
Product / Packaging disposal	 Containers may still present a chemical hazard/ danger when empty. Return to supplier for reuse/ recycling if possible. Otherwise: If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill. Where possible retain label warnings and SDS and observe all notices pertaining to the product. DO NOT allow wash water from cleaning or process equipment to enter drains. It may be necessary to collect all wash water for treatment before disposal. In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first. Where in doubt contact the responsible authority. Recycle wherever possible or consult manufacturer for recycling options. Consult State Land Waste Authority for disposal. Bury or incinerate residue at an approved site. Recycle containers if possible, or dispose of in an authorised landfill.

SECTION 14 Transport information

Labels Required	
Marine Pollutant	
HAZCHEM	•3Z
Land transport (ADG)	

14.1. UN number or ID number	3082	3082		
14.2. UN proper shipping name	ENVIRONMENTALLY	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)		
14.3. Transport hazard class(es)	Class Subsidiary Hazard	9 Not Applicable		
14.4. Packing group	II			
14.5. Environmental hazard	Environmentally hazardous			
14.6. Special precautions for user	Special provisions Limited quantity	274 331 335 375 AU01 5 L		

Environmentally Hazardous Substances meeting the descriptions of UN 3077 or UN 3082 are not subject to this Code when transported by road or rail in;

(a) packagings;(b) IBCs; or

(c) any other receptacle not exceeding 500 kg(L).
 Australian Special Provisions (SP AU01) - ADG Code 7th Ed.

Air transport (ICAO-IATA / DGR)

14.1. UN number	3082			
14.2. UN proper shipping name	Environmentally hazardous substance, liquid, n.o.s. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)			
	ICAO/IATA Class	9		
14.3. Transport hazard class(es)	ICAO / IATA Subsidiary Hazard	Not Applicable		
0,000(00)	ERG Code	9L		
14.4. Packing group	III			
14.5. Environmental hazard	Environmentally hazardous			
	Special provisions		A97 A158 A197 A215	
	Cargo Only Packing Instructions		964	
14.6. Special precautions for user	Cargo Only Maximum Qty / Pack		450 L	
	Passenger and Cargo Packing Instructions		964	
	Passenger and Cargo Maximum Qty / Pack		450 L	
	Passenger and Cargo Limited Quantity Packing Instructions		Y964	
	Passenger and Cargo Limited Maximum Qty / Pack		30 kg G	

Sea transport (IMDG-Code / GGVSee)

14.1. UN number	3082			
14.2. UN proper shipping name	ENVIRONMENTALLY	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains copper sulfate, pentahydrate and zinc sulfate heptahydrate)		
14.3. Transport hazard class(es)	IMDG Class IMDG Subsidiary Ha	9 azard Not Applicable		
14.4. Packing group	III			
14.5 Environmental hazard	Marine Pollutant			
14.6. Special precautions for user	EMS Number Special provisions Limited Quantities	F-A , S-F 274 335 969 5 L		

14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
manganese sulfate, hydrate	Not Available
zinc sulfate heptahydrate	Not Available
phosphoric acid	Not Available
copper sulfate, pentahydrate	Not Available
ferrous sulfate heptahydrate	Not Available
urea	Not Available
sodium molybdate	Not Available

14.7.3. Transport in bulk in accordance with the IGC Code

Product name	Ship Type
manganese sulfate, hydrate	Not Available
zinc sulfate heptahydrate	Not Available
phosphoric acid	Not Available
copper sulfate, pentahydrate	Not Available
ferrous sulfate heptahydrate	Not Available
urea	Not Available
sodium molybdate	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

manganese sulfate, hydrate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals
Australian Inventory of Industrial Chemicals (AIIC)

zinc sulfate heptahydrate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 6 Australian Inventory of Industrial Chemicals (AIIC)

phosphoric acid is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 5 Australian Inventory of Industrial Chemicals (AIIC)

copper sulfate, pentahydrate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 5 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 6 Australia Inventory of Industrial Chemicals (AIIC)

ferrous sulfate heptahydrate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 2 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 5 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 6 Australia Inventory of Industrial Chemicals (AIIC)

urea is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

sodium molybdate is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

Additional Regulatory Information

Not Applicable

National Inventory Status

National Inventory	Status	
Australia - AIIC / Australia Non- Industrial Use	Yes	
Canada - DSL		
Canada - NDSL	No (manganese sulfate, hydrate; zinc sulfate heptahydrate; phosphoric acid; copper sulfate, pentahydrate; ferrous sulfate heptahydrate; urea; sodium molybdate)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	No (ferrous sulfate heptahydrate)	

National Inventory	Status	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

SECTION 16 Other information

Revision Date	14/03/2024
Initial Date	07/03/2024

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
- TEEL: Temporary Emergency Exposure Limit.
- IDLH: Immediately Dangerous to Life or Health Concentrations
- ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level
- LOAEL: Lowest Observed Adverse Effect Level
- TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value
- BCF: BioConcentration Factors
- BEI: Biological Exposure Index
- DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration
- AIIC: Australian Inventory of Industrial Chemicals
- DSL: Domestic Substances List
- NDSL: Non-Domestic Substances List
- IECSC: Inventory of Existing Chemical Substance in China
- EINECS: European INventory of Existing Commercial chemical Substances
- ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
 ENCS: Existing and New Chemical Substances Inventory
- KECI: Korea Existing Chemicals Inventory
- NZIoC: New Zealand Inventory of Chemicals
- PICCS: Philippine Inventory of Chemicals and Chemical Substances
- TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory
- INSQ: Inventario Nacional de Sustancias Químicas
- NCI: National Chemical Inventory
- FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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